

Dust formation in AGB stars

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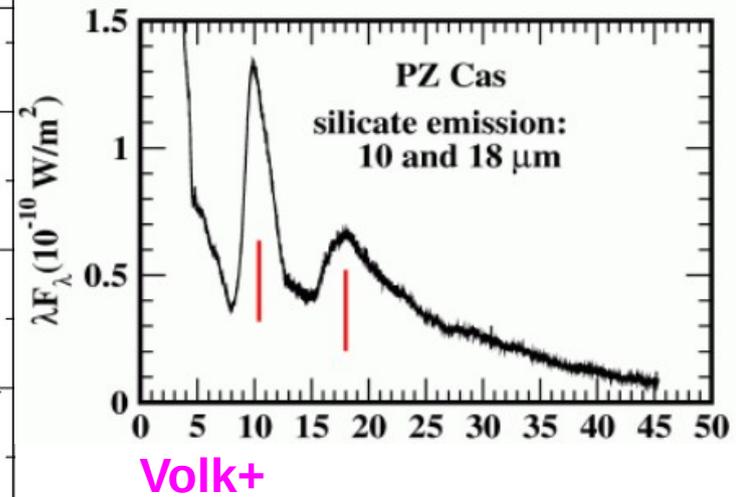
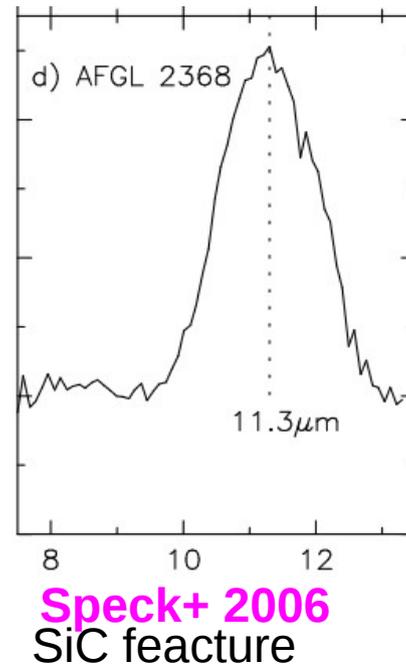
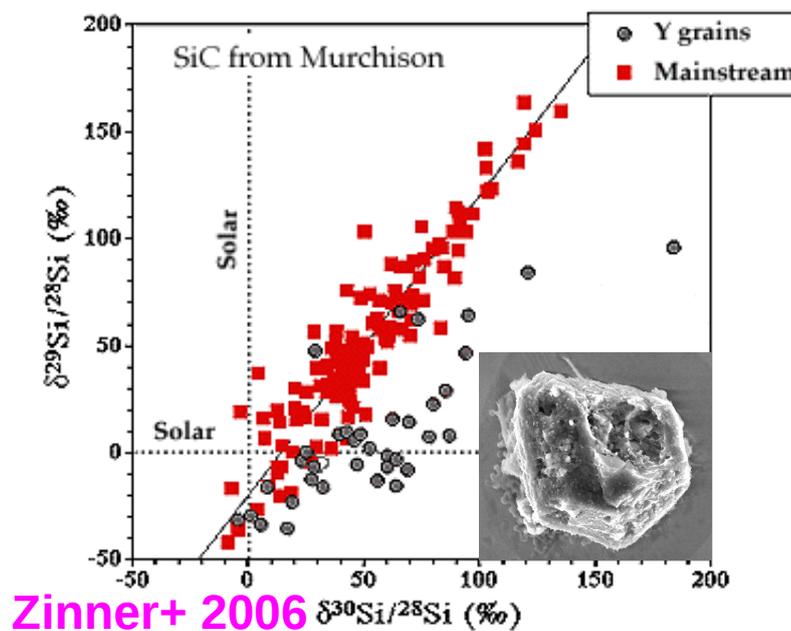
Luciano Piersanti

Stefan T. Bromley

Isabelle Cherchneff & Arkaprabha Sarangi

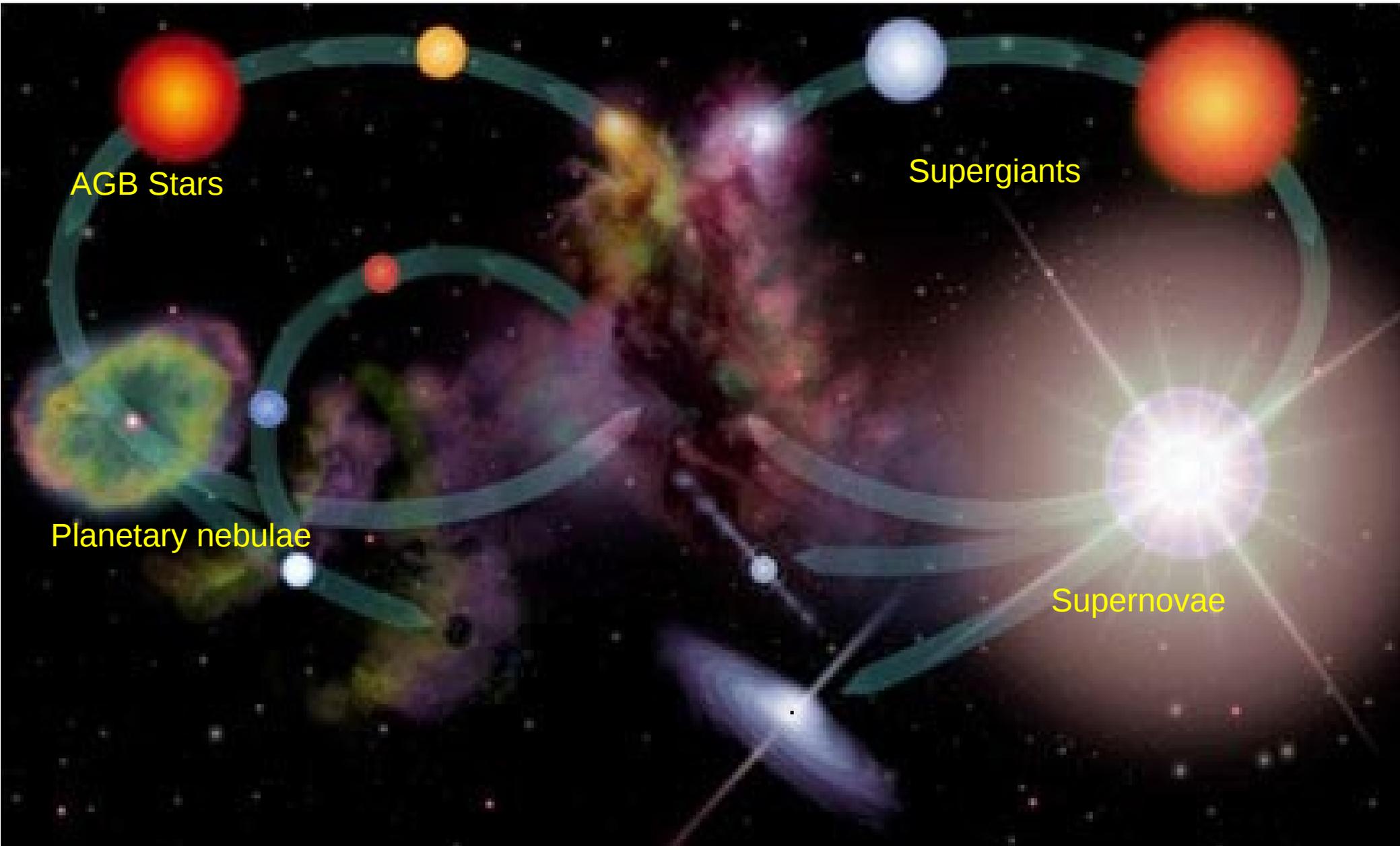
Evidence for dust

- In presolar meteoritic grains with particular isotopic signatures



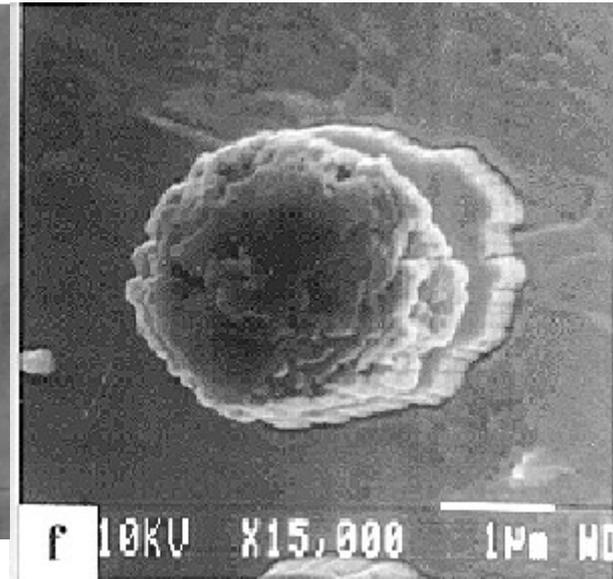
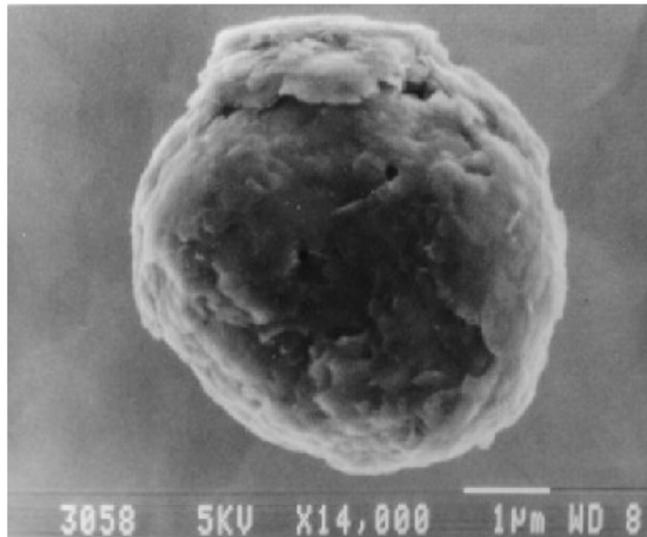
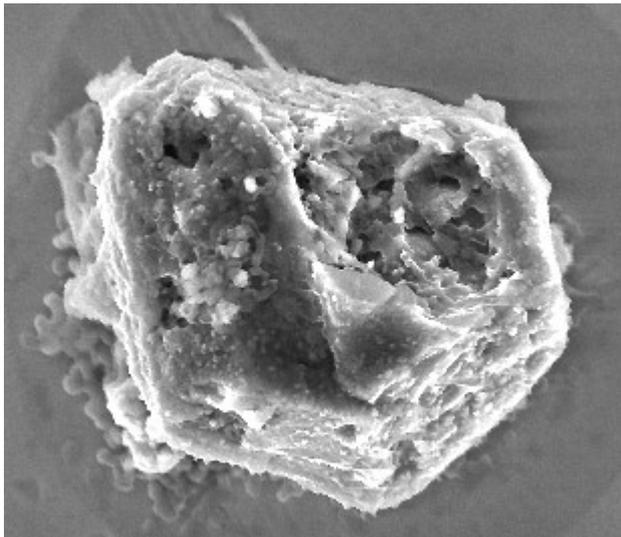
- Spectroscopic features
- Depletion of atoms and molecules (indirect)

Stellar sources of dust



Types of dust

- C-rich dust: (hydrogenated) amorphous carbon, graphite, SiC
- O-rich dust: silicates (olivine, pyroxene) , metal oxides (Al_2O_3 , $\text{TiO}_{(2)}$, $\text{SiO}_{(2)}$, FeO)
- Pure Metals: Fe, Mg, Al

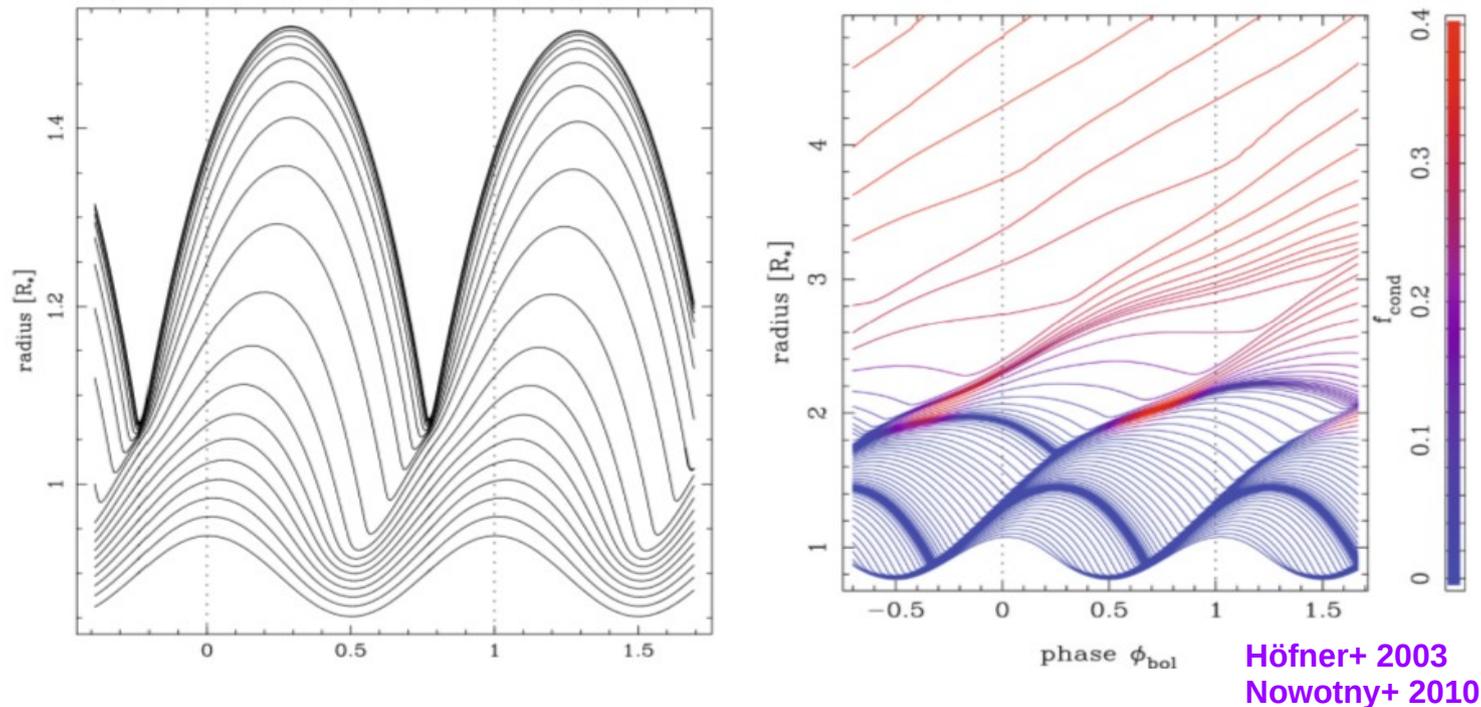


Amari+ 1994

Conditions for dust formation

- Low temperatures (< 2000 K) to afford condensation
- Sufficient high densities ($\sim 10^{10} - 10^{16}$ cm $^{-3}$) to allow for coalescence
- Availability of atoms, molecules & clusters for the specific condensate
- Sufficient time for dust clusters and grains to grow (time scales)
 - such conditions prevail in the inner envelopes of AGB stars

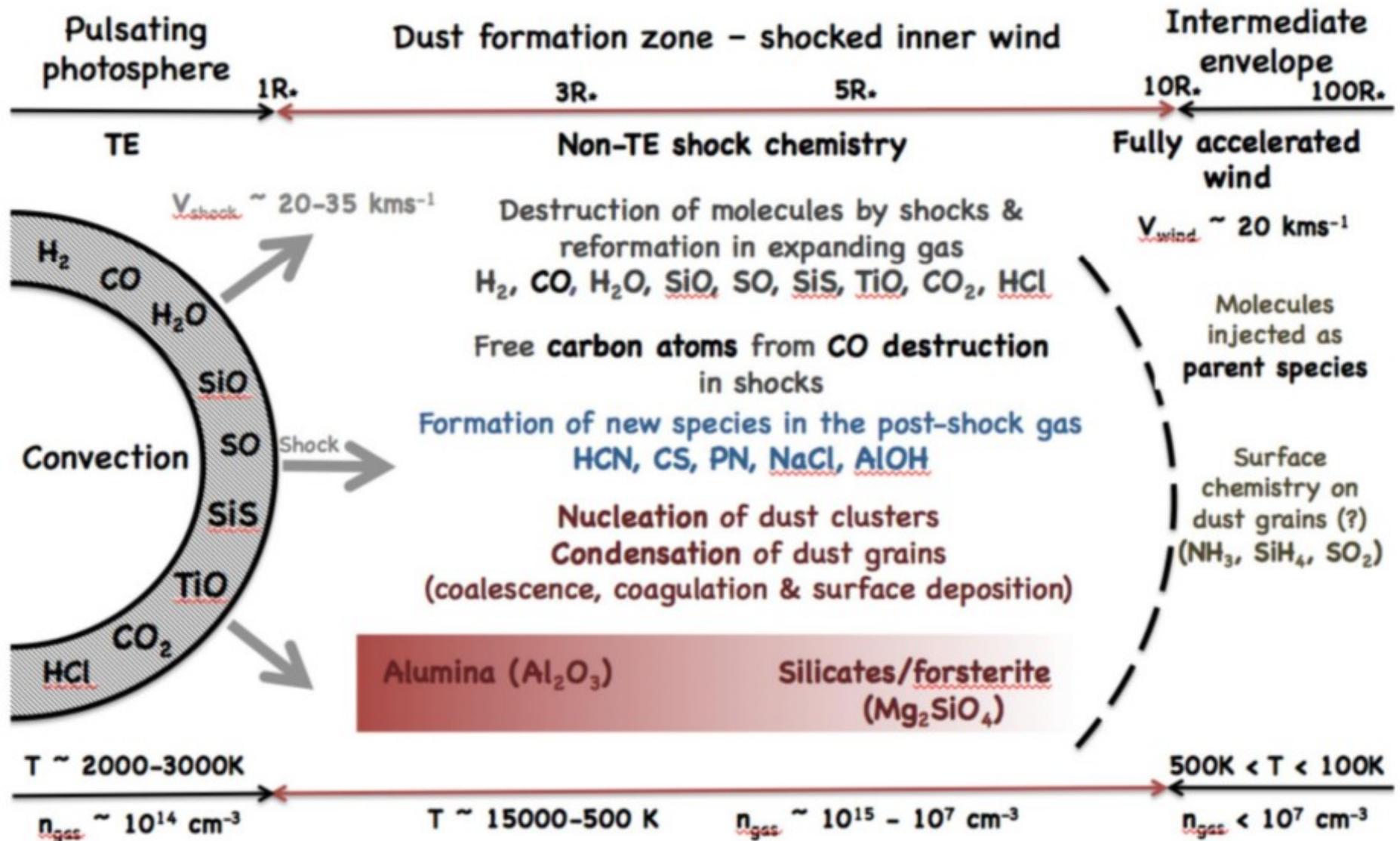
Pulsational shocks



- Periodic pulsations compress, heat and accelerate the ambient medium
- dense molecular layers form in the cooling post-shock gas
 - favourable conditions for dust formation

Circumstellar envelope

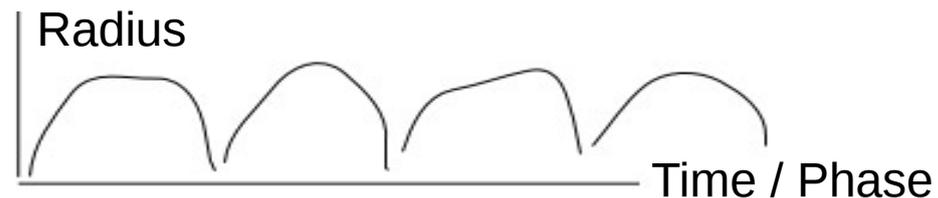
Gobrecht et al.: Dust formation in IK Tau



IK Tau: circumstellar model

- Periodic pulsation model of the extended atmosphere of an O-rich AGB + chemical-kinetic network with 100 species and 440 processes including dust cluster formation

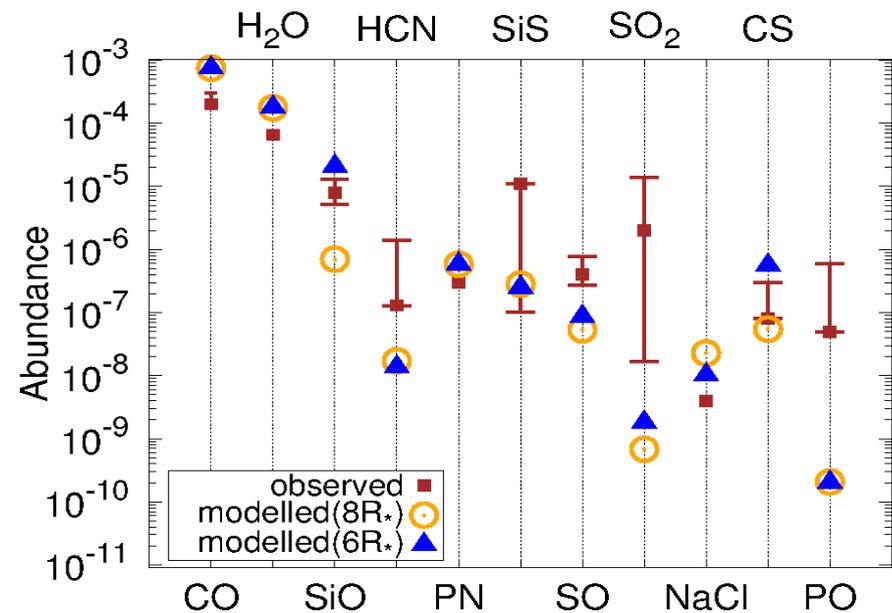
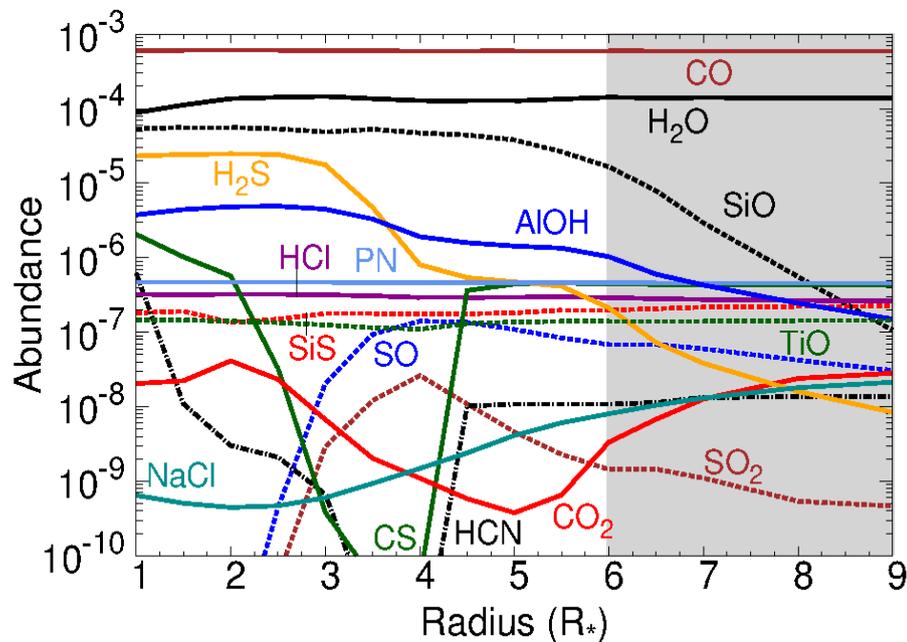
Star		
$M_{\star,ZAMS}$	$1.5 M_{\odot}$	1
R_{\star}	2.5×10^{13} cm	2
C/O	0.75	3
P	470 days	4
T_{eff}	2200 K	2
Wind		
r_s	$1 R_{\star}$	3
V_s	32 km s^{-1}	3
\dot{M}	$8 \times 10^{-6} M_{\odot} \text{ yr}^{-1}$	2
v_{esc}	17.7 km s^{-1}	2
γ	0.98	
α	-0.6	3
$n_{gas}(r_s)$	$3.62 \times 10^{14} \text{ cm}^{-3}$	3
$T_{gas}(r_s)$	2200 K	-



Reaction type	Reaction	formulation	Gas conditions
<i>Unimolecular</i>			
Thermal decomposition	AB	$\rightarrow A + B$	High T
<i>Bimolecular</i>			
Neutral-exchange	AB + C	$\rightarrow A + CB$	T dependent
Collisional dissociation	AB + M	$\rightarrow A + B + M$	High T
Radiative association	A + B	$\rightarrow AB + \gamma$	T independent
<i>Termolecular</i>			
Termolecular formation	A + B + M	$\rightarrow AB + M$	(Very) high n

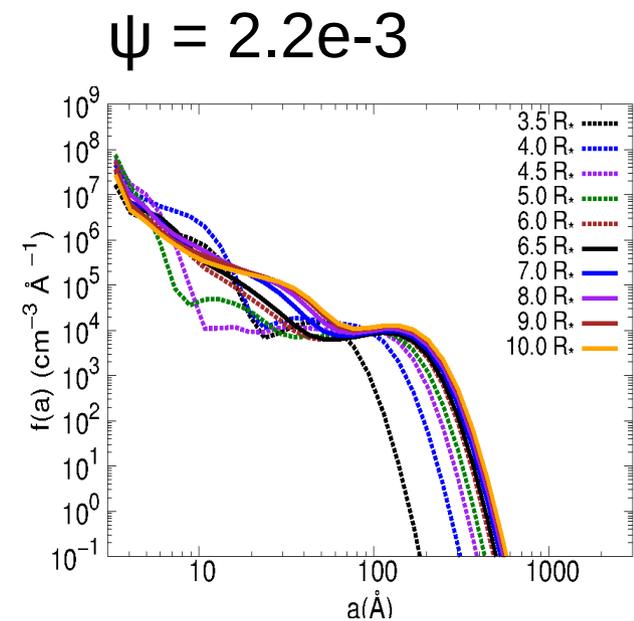
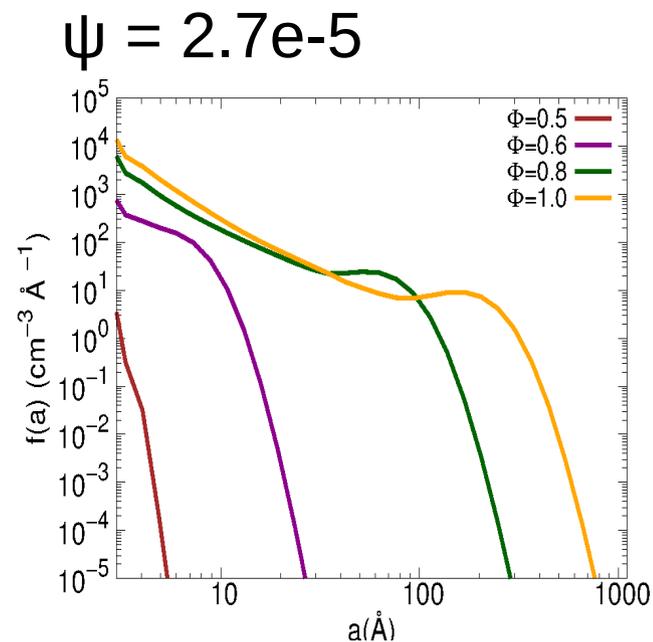
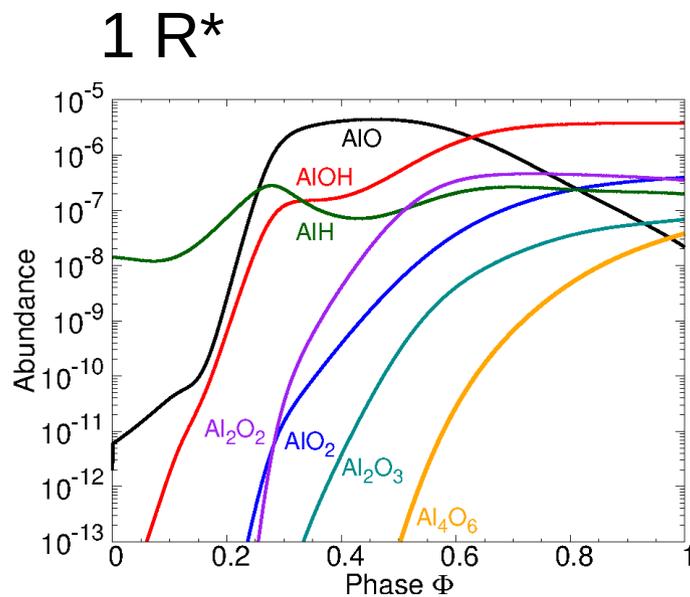
IK Tau: results on molecules

- Non-equilibrium: Carbon-bearing species like HCN, CS or CO₂ form in the inner envelope
- Modelled abundances at 6-8 R_{*} agree well with observations



IK Tau: results on clusters and dust

- Clusters of alumina (Al_2O_3) form and condense close to the star in the cooling post-shock gas
- Forsterite grains (Mg_2SiO_4) form at larger radii ($4R_*$) in greater amounts, but lower size



Conclusions (part I)

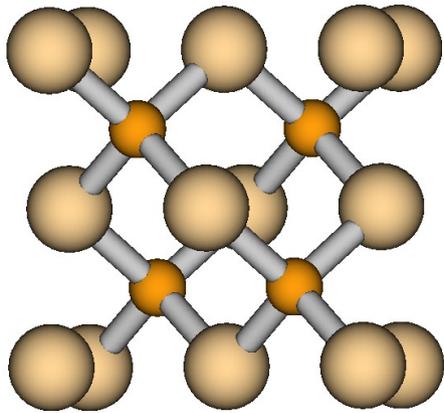
- Molecular abundances including non-equilibrium species HCN, CS and CO₂ well reproduced, expect NH₃, PO and (SO₂)
- Large grains of alumina dust form close to the star that is too low to exclusively explain the observed mass loss rate / dust-to-gas ratio
- Forsterite grains form around 4 R_{*} in amounts comparable to observations, although their mean size rather low

Dust Clusters

- Link between molecules and dust grains:
e.g. Al_4O_6 , $\text{Mg}_4\text{Si}_2\text{O}_8$, $\text{Si}_{16}\text{C}_{16}$
- Many body-problem: $H\psi(r) = E\psi(r)$
 - no exact solution
 - parallelized supercomputers required to perform full quantum calculations
 - structures, energetics, reactivity are not well characterized

Silicon carbide (SiC) clusters

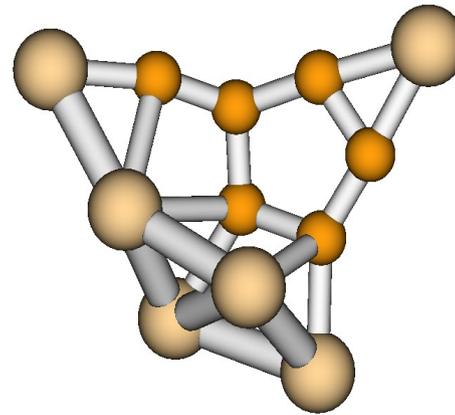
Crystalline bulk



versus

clusters

4.4 Å



Clusters: Quantum effects, surface-to-volume ratio, unsaturated bonds

Candidate structure search by Molecular dynamic & Monte-Carlo simulations

→ subsequent refinement on the quantum DFT level of theory

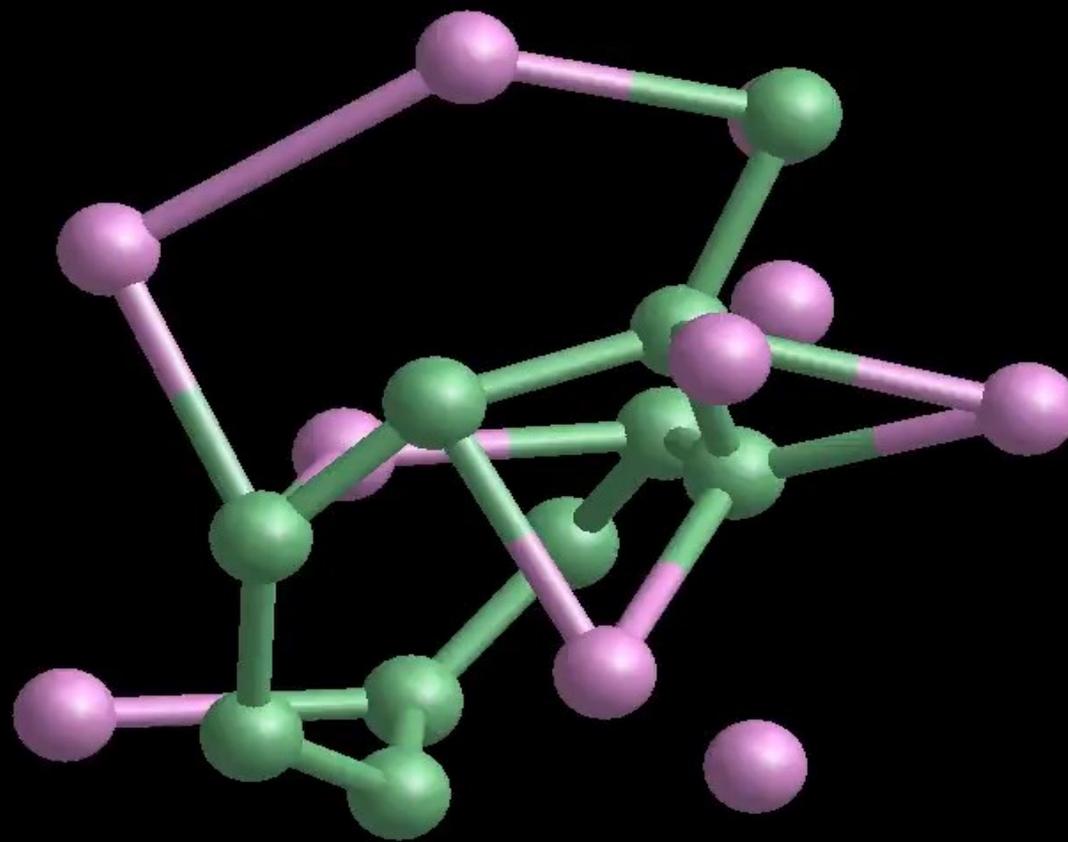
SiC: Molecular dynamics

- Tersoff potential:

Tersoff 1989

$$V_{ij} = f_C(r_{ij}) [V_{rep}(r_{ij}) + b_{ij}(\theta_{ijk})V_{att}(r_{ij})]$$

- accounts for covalent character of clusters
- segregation of Si and C atoms, C₅ and C₆ ring formation
- Simulated annealing: High-temperature (2000-3000 K) molecular dynamics and subsequent decrease of temperature with the program GULP (General Utility Lattice Program)



Monte-Carlo candidate structures

- Buckingham pair potential:

$$\Phi_{12}(r) = A \exp(-Br) - \frac{C}{r^6} + \frac{q_1 q_2}{4\pi\epsilon_0 r}$$

Buckingham 1938

→ account for ionic character of clusters

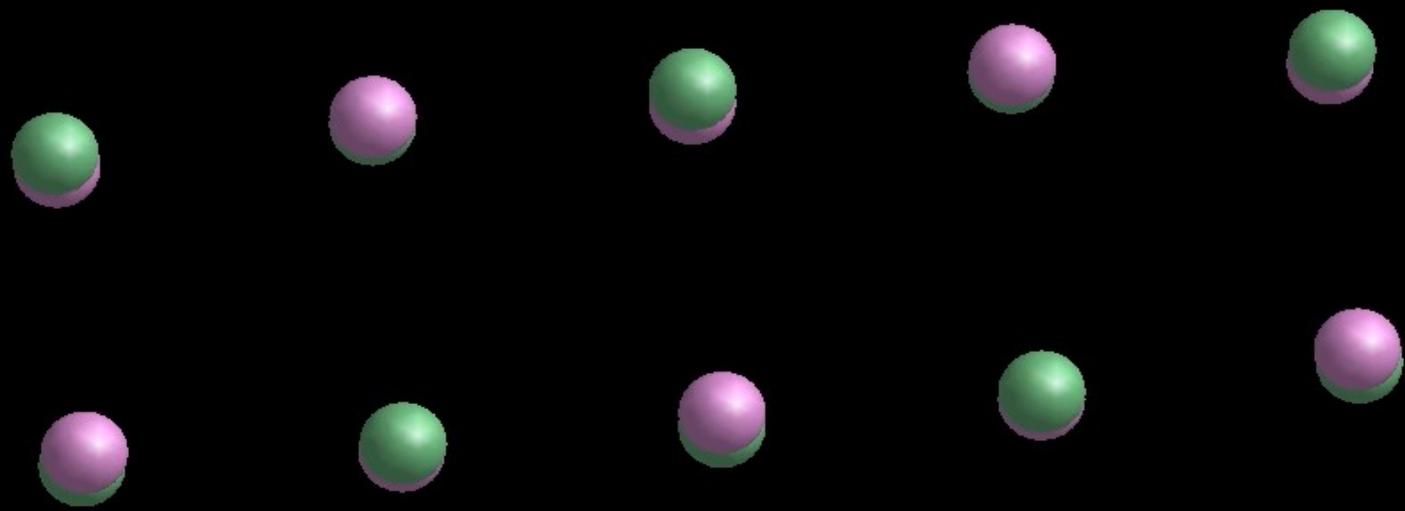
→ alternating Si-C bondings

- ZnS (zinkblende) analogy, charge transfers between C and Si ions as large as 2.5 e

Watkins 2009

- Monte-Carlo exploration of the potential landscape with the Basin-Hopping algorithm

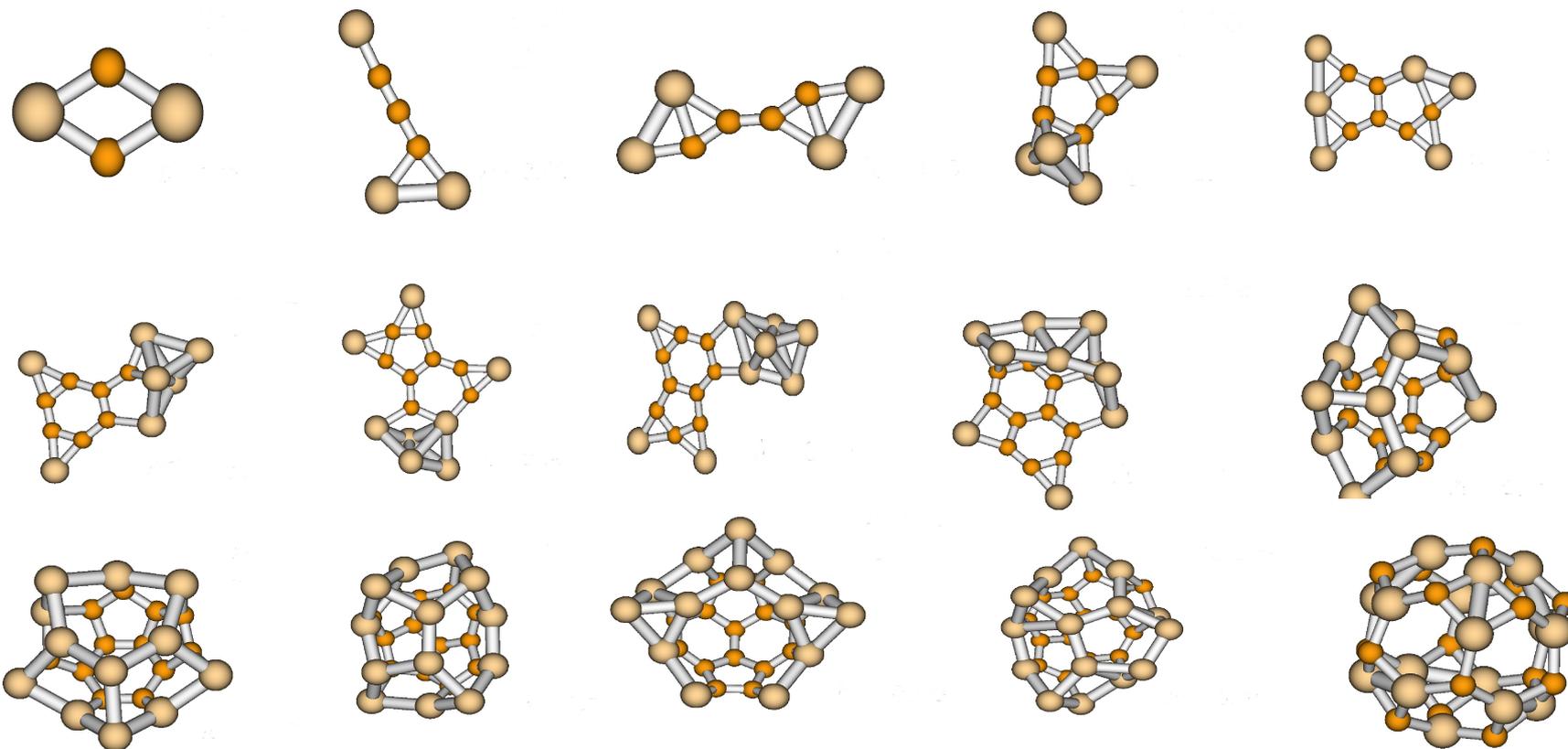
Wales 1997
Bromley 2005



SiC clusters: preliminary results

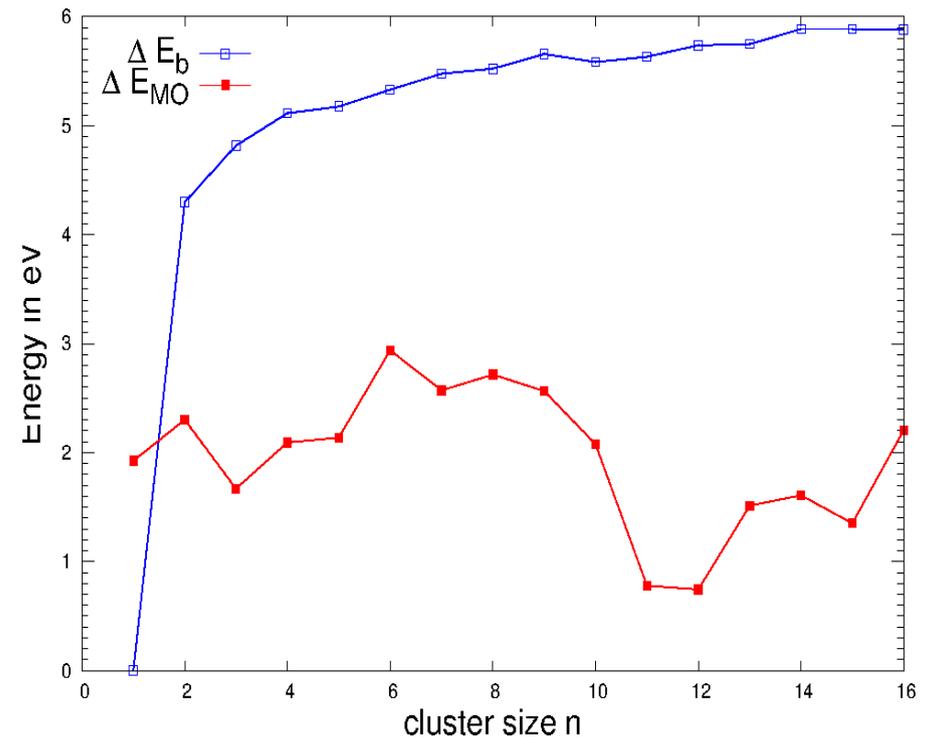
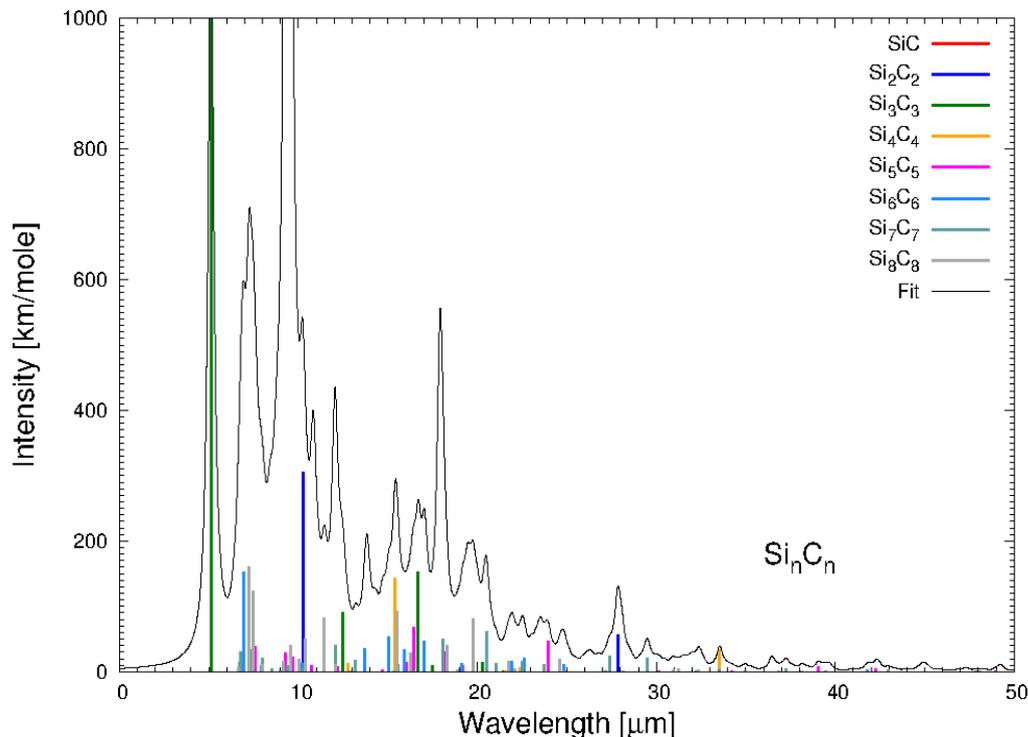
- Most stable Si_nC_n structures exhibit segregation of C atoms forming chains, C_5 and C_6 rings and cage-like configurations with increasing size

Gobrecht in preparation



SiC clusters: preliminary results II

- DFT calculations → Potential energy, HOMO-LUMO gap, Vibrational IR spectra, theoretical reaction rate determination



Conclusions (part II)

- We found new, stable $(\text{SiC})_n$ with $n \leq 16$ clusters with the lowest potential energies
- Segregated clusters with C_5 / C_6 rings dominate the SiC cluster chemistry for $n \leq 16$
- However, for $n=12$ and $n=16$ very stable clusters with alternating Si-C bonds and bulk-like properties are found
- These clusters indicate the sizes / length scales, where the transition of quantum-effect dominated clusters and bulk-SiC (found in meteoritic inclusions) may occur

Still to be finalized ...

- Comparison with observations (e.g. 11.3 μm feature)
- Theoretical rate (temperature & density dependent)
- Formation routes via Si_nC_m , $m \neq n$ (SiC_2 and Si_2C observed)

THANK YOU FOR YOUR ATTENTION!